Studies on Fungal Products. Part 15.¹ Isolation and Structure Determination of Arugosin E from *Aspergillus silvaticus* and Cycloisoemericellin from *Emericella striata*

Nobuo Kawahara, Koohei Nozawa, Shoichi Nakajima, and Ken-ichi Kawai* Faculty of Pharmaceutical Sciences, Hoshi University, Ebara 2-4-41, Shinagawa-ku, Tokyo 142, Japan

Together with arugosins A (3), B (4), and C (5), a novel compound related to the arugosins, designated arugosin E (1), was isolated from the mycelial extract of *Aspergillus silvaticus*. A new compound designated cycloisoemericellin (2) was also isolated along with arugosins A (3) and B (4) from the mycelial extract of *Emericella striata*. The structures of compounds (1) and (2) were determined on the basis of spectroscopic investigation and chemical correlation with the known compounds (3), (4), and (6). An X-ray crystal analysis of arugosin E (1) was carried out.

In the previous paper,² we reported the isolation of two novel dioxopiperazine derivatives, dithiosilvatin and silvathione, from *Aspergillus silvaticus* Fennell & Raper, strain IFO 8173. In the search for other metabolites from the above fungus, a new compound designated arugosin E (1) was isolated along with dithiosilvatin and arugosins A (3), B (4), and C (5) from the mycelial extract. Compounds (3), (4), and (5) are metabolites of *Emericella rugulosa* (Thom & Raper) C. R. Benjamin (anamorph: *A. rugulosus* Thom & Raper)^{3,4} and *Emericella variecolor* Berkley & Broome (anamorph: *A. variecolor* Thom & Raper).^{5,6} Cycloisoemericellin (2) was isolated with arugosins A

(3) and B (4) from the non-polar fraction of the acetone extract of *Emericella striata* (Rai, Tewari & Mukerji) Malloch & Cain (anamorph: *A. striatus* Rai, Tewari & Mukerji), strain 80-NE-22. The structural elucidation of compounds (1) and (2) is reported in this paper.

Results and Discussion

Arugosin E (1), m.p. 159—161 °C, gave a molecular ion at m/z422 by electron impact (e.i.) mass spectrometry, and elemental analysis confirmed the molecular formula as $C_{25}H_{26}O_6$. The ¹H













(5)



Table 1. ¹H. N.m.r. chemical shifts of arugosin E (1), cycloisoemericellin (2), and their derivatives in $CDCl_3$

Carbon ^a					
No.	(1)	(2)	(3) and (4)	(5)	(6)
1-OH			(13.44, 12.83)	13.87	12.50
2-H			(, 6.66)		6.70
3-H	7.01	7.25	(7.27, 7.36)	7.28	7.40
4-H	6.49	6.86	(6.44,)	6.40	
4a-OH	12.43				
5-H	6.99	7.21	6.88	6.81	7.24
5a-OH	5.40		(10.88, 11.15)	10.70	
6-Me	2.36	2.42	2.35	2.24	2.42
11-H	10.19	5.02	6.54	5.08	5.04
11-OH		4.98	5.00		4.45
1′-H	6.12 ^{<i>b</i>}	6.33 ^b	3.31	3.49	3.46
2′-H	5.26 ^b	5.64 <i>^b</i>	5.28	5.30	5.30
4′-H	0.94	1.57	1.72	1.76 <i>^b</i>	1.73 ^b
5′-H	0.94	1.57	1.72	1.73 <i>*</i>	1.68 ^b
1″-H	4.39	4.45	4.41	4.27	4.42
				4.38	
2″-H	5.51	5.61	5.53	2.35	5.60
3″-OH				1.55	
4″-H	1.77	1.81 °	1.82 ^b	1.26°	1.85 ^b
5″-H	1.63	1.73°	1.72*	1.32°	1.81 ^b

^{*a*} Numberings of the related compounds correspond to those of compound (6). bc Assignments may be reversed.

 Table 2. Final atomic positional parameters with estimated standard deviations in parentheses

Atom	x	У	Ζ
O(1)	0.983 3(4)	0.203 5(2)	0.290 1(3)
O(2)	0.830 3(5)	0.430 9(2)	0.445 1(3)
O(3)	0.644 4(4)	0.337 6(2)	0.376 1(3)
O(4)	0.558 9(5)	0.030 9(2)	0.298 1(3)
O(5)	0.718 6(5)	0.315 1(2)	0.162 4(3)
O(6)	0.728 4(5)	0.186 9(3)	0.473 3(3)
C(2)	1.086 3(7)	0.172 0(4)	0.244 2(5)
C(3)	1.243 8(8)	0.183 4(5)	0.264 0(6)
C(4)	1.279 0(7)	0.240 6(5)	0.319 0(6)
C(4a)	1.164 5(7)	0.292 5(4)	0.349 1(4)
C(5)	1.197 2(7)	0.361 7(4)	0.397 5(4)
C(6)	1.084 5(7)	0.407 6(4)	0.430 0(4)
C(7)	0.936 9(7)	0.384 8(3)	0.412 4(4)
C(8)	0.900 5(6)	0.316 2(3)	0.363 5(4)
C(8a)	1.016 8(6)	0.272 0(3)	0.331 6(4)
C(9)	1.040 8(9)	0.087 9(4)	0.213 4(7)
C(10)	1.061 0(10)	0.220 1(6)	0.132 9(5)
C(11)	0.744 2(6)	0.297 1(3)	0.346 3(3)
C(1')	0.659 6(6)	0.157 2(3)	0.320 1(4)
C(2')	0.605 3(6)	0.099 6(3)	0.261 5(4)
C(3')	0.586 4(6)	0.113 9(3)	0.165 5(4)
C(4′)	0.624 3(7)	0.185 9(4)	0.133 4(4)
C(5')	0.679 0(6)	0.243 9(3)	0.192 0(4)
C(6')	0.694 3(5)	0.230 3(3)	0.286 6(3)
C(7′)	0.681 6(7)	0.140 1(4)	0.418 8(4)
C(8′)	0.514 9(9)	0.053 8(4)	0.101 6(5)
C(9′)	0.644 7(11)	-0.039 2(4)	0.280 2(6)
C(10')	0.772 2(13)	-0.049 0(5)	0.347 3(7)
C(11')	0.767 0(11)	-0.0863(5)	0.427 6(6)
C(12')	0.901 2(14)	-0.097 4(7)	0.493 8(9)
C(13')	0.634 1(15)	-0.122 6(11)	0.462 1(10)

n.m.r. signals of compound (1) were similar to those of (3) or (4), with the exception of (i) the appearance of the aldehyde proton at δ 10.19 in (1) instead of the hemiacetal proton at δ 6.54 in (3) and (4), and (ii) the disappearance of the signals assigned to one of the 3-methylbut-2-enyl groups in (3) and (4) (Table 1). The signals observed at δ 5.26 (1 H, d) and 6.12 (1 H, d) in (1) were

Table 3. Bond lengths (Å) for arugosin E (1) with estimated standard deviations in parentheses

O(1) - C(2)	1.477(9)	O(1)-C(8a)	1.358(7)
O(2) - C(7)	1.355(8)	O(3) - C(11)	1.239(7)
O(4) - C(2')	1.376(7)	O(4)–C(9')	1.470(11)
O(5) - C(5')	1.358(8)	O(6)-C(7')	1.198(8)
C(2) - C(3)	1.524(12)	C(2) - C(9)	1.517(13)
C(2)-C(10)	1.579(13)	C(3) - C(4)	1.304(12)
C(4)-C(4a)	1.454(11)	C(4a)-C(5)	1.413(9)
C(4a)-C(8a)	1.393(9)	C(5) - C(6)	1.394(10)
C(6)-C(7)	1.404(9)	C(7) - C(8)	1.415(8)
C(8) - C(8a)	1.398(8)	C(8) - C(11)	1.461(8)
C(11) - C(6')	1.503(8)	C(1') - C(2')	1.386(8)
C(1')-C(6')	1.395(8)	C(1') - C(7')	1.476(8)
C(2')-C(3')	1.426(8)	C(3')-C(4')	1.378(9)
C(3') - C(8')	1.520(10)	C(4') - C(5')	1.392(9)
C(5')-C(6')	1.402(8)	C(9')-C(10')	1.485(15)
C(10')-C(11')	1.341(16)	C(11')-C(12')	1.526(17)
C(11')-C(13')	1.467(23)		



Figure. Perspective view of the crystal structure of arugosin E (1) with thermal ellipsoids at 50% probability

assigned as the vicinal olefinic protons, and those at $\delta 0.94$ (6 H, s) in (1) as the two aliphatic methyl groups. The above results suggested that the 3-methylbut-2-enyl group in (3) or (4) was converted into a 2,2-dimethylchromene skeleton in (1). 3-Methylbut-2-enyl groups usually undergo cyclisation with a hydroxy group at the *ortho*-position of a benzene ring on treatment with 2,3-dichloro-5,6-dicyano-*p*-benzoquinone (DDQ)⁷ to give a 2,2-dimethylchlromene skeleton. When arugosins A (3) and B (4) were treated with DDQ, their hemiacetal bonds were cleaved and oxidative cyclization gave compound (1). The structure of arugosin E was therefore assumed to be that in (1).

In order to determine the exact structure of arugosin E (1), an X-ray structure analysis was undertaken (see Figure). Most of the bond lengths and angles were as expected (Tables 3 and 4). An intermolecular hydrogen bond of 2.769 Å was apparent between O(5) and O(6). The molecules of (1) are packed closely together in the c direction through hydrogen bonding. An intramolecular hydrogen bond of 2.504 Å was Table 4. Bond angles $(^{\circ})$ for arugosin E (1) with estimated standard deviations in parentheses

C(2)-O(1)-C(8a)	118.6(5)	C(2')-O(4)-C(9')	118.0(6)
O(1)-C(2)-C(3)	108.1(6)	O(1)-C(2)-C(9)	104.1(6)
O(1)-C(2)-C(10)	106.7(6)	C(3)-C(2)-C(9)	114.0(7)
C(3)-C(2)-C(10)	110.4(7)	C(9)-C(2)-C(10)	113.0(7)
C(2)-C(3)-C(4)	121.9(8)	C(3)-C(4)-C(4a)	119.9(8)
C(4)-C(4a)-C(5)	122.5(6)	C(4)-C(4a)-C(8a)	118.6(6)
C(5)-C(4a)-C(8a)	118.8(6)	C(4a)-C(5)-C(6)	120.9(6)
C(5)-C(6)-C(7)	118.9(6)	O(2)-C(7)-C(6)	117.2(6)
O(2)-C(7)-C(8)	121.2(5)	C(6)-C(7)-C(8)	121.6(6)
C(7)-C(8)-C(8a)	117.7(5)	C(7)-C(8)-C(11)	118.4(5)
C(8a)-C(8)-C(11)	123.9(5)	O(1)-C(8a)-C(4a)	119.6(5)
O(1)-C(8a)-C(8)	118.0(5)	C(4a) - C(8a) - C(8)	122.1(5)
O(3)-C(11)-C(8)	121.8(5)	O(3)-C(11)-C(6')	115.8(5)
C(8)-C(11)-C(6')	122.2(5)	C(2')-C(1')-C(6')	120.8(5)
C(2')-C(1')-C(7')	119.0(5)	C(6')-C(1')-C(7')	120.1(5)
O(4)-C(2')-C(1')	119.1(5)	O(4)-C(2')-C(3')	120.5(5)
C(1')-C(2')-C(3')	120.2(5)	C(2')-C(3')-C(4')	118.1(5)
C(2')-C(3')-C(8')	120.8(6)	C(4')-C(3')-C(8')	120.9(6)
C(3')-C(4')-C(5')	121.8(6)	O(5)-C(5')-C(4')	123.2(5)
O(5)-C(5')-C(6')	116.8(5)	C(4')-C(5')-C(6')	120.0(5)
C(11)-C(6')-C(1')	123.9(5)	C(11)-C(6')-C(5')	117.1(5)
C(1')-C(6')-C(5')	119.0(5)	O(6)-C(7')-C(1')	122.7(6)
O(4)-C(9')-C(10')	112.1(8)	C(9')-C(10')-C(11')	124.9(10)
C(10')-C(11')-C(12')	123.6(10)	C(10')-C(11')-C(13')	124.8(12)
C(12')-C(11')-C(13')	111.6(12)		

Table 5. Selected torsion angles (°) with estimated standard deviations in parentheses

O(3)-C(11)-C(8)-C(7)	-2.7(0.8)
O(3)-C(11)-C(8)-C(8a)	179.8(0.5)
O(3)-C(11)-C(6')-C(5')	91.1(0.6)
O(3)-C(11)-C(6')-C(1')	-87.3(0.7)
O(6)-C(7')-C(1')-C(6')	0.4(0.9)
O(6)-C(7')-C(1')-C(2')	179.8(0.6)

apparent between O(2) and O(3), whereas the distance of 3.248 Å found between O(3) and O(5) suggested no hydrogen bonding. The torsion angles around the ketones are given in Table 5. The dihedral angles between the least-squares planes formed by the chromane ring and the ketone, and by the benzene ring and the ketone are 4.5° and 93.0°, respectively. All the above results confirmed that the planes formed by the chromanylketone and the benzene ring in compound (1) were almost perpendicular. This was also shown in the ¹H n.m.r. spectra—the signal at δ 12.43 being assigned as the phenolic hydroxy which was hydrogen-bonded to the ketone, and the signal at δ 5.40 being assigned as the non hydrogen-bonded phenolic hydroxy of the benzene ring. The ketone signal observed at $\delta_{\rm C}$ 200.44 also indicated that there was no conjugation between chromanylketone and the benzene ring in (1). The assignments of ${}^{13}C$ n.m.r. signals of arugosin E (1) are given in Table 6.

Cycloisoemericellin (2), m.p. 136–138 °C, gave a molecular ion at m/z 406 by e.i. m.s. and elemental analysis confirmed the molecular formula as $C_{25}H_{26}O_5$. The u.v. maxima at 238, 280, 317, and 380 nm in (2) suggested the presence of a xanthone nucleus as in emericellin (6) (u.v. maxima at 237, 271, 294, and 384 nm). The ¹H n.m.r. signals observed at δ 1.73 (3 H), 1.81 (3 H), 5.61 (1 H, br t), and 4.45 (2 H, br d) were assigned to the 3-methylbut-2-enyloxy group in (2). The ¹H (Table 1) and ¹³C (Table 6) n.m.r. signals of (2) were similar to those of (6), with the exception of signals observed at δ 1.57 (6 H, s), 5.64 (1 H, d), and 6.33 (1 H, d) in the ¹H n.m.r. spectrum and at δ_C 28.01 (Qm), 77.88 (Sm), 129.42 (Dm), and 121.61 (Dd) in the ¹³C n.m.r. **Table 6.** 13 C N.m.r. chemical shifts of arugosin E (1), cycloisoemericellin (2), and emericellin (6) in CDCl₃

Carbon ^{<i>a</i>}			
No.	(1)	(2)	(3)
1	154.42 (Sdd)	154.39 (Sdd)	154.08 (S)
2	112.72 (Sddd)	116.91 (Sddd)	110.14 (D)
3	133.83 (Dd)	131.77 (Dd)	136.97 (D)
4	109.70 (Dd)	108.33 (D)	109.20 (S)
4a	163.62 (Sm)	156.42 (Sdd)	160.09 (S)
5	125.06 (Sq)	118.84 (Dq)	118.94 (D)
6	134.49 (Sq)	140.50 (Sq)	142.68 (S)
6-Me	15.88 (Qd)	17.47 (Qd)	17.82 (Q) ^b
7	154.27 (Sm)	152.38 (Sm)	152.79 (S) ^c
8	128.75 (Sd)	134.58 (S br,t)	134.45 (S)
8a	128.45 (Sm)	120.73 (Sm)	121.81 (S)
9	200.44 (S)	179.10 (S)	184.75 (S)
9a	112.05 (S br,dd)	112.86 (Sd)	118.11 (S)
10a	148.15 (Sd)	152.84 (Sd)	152.95 (S) ^c
11	189.43 (D)	57.20 (Td)	57.23 (T)
1′	121.92 (Dd)	121.61 (Dd)	27.58 (T)
2′	126.43 (Dm)	129.42 (Dm)	119.47 (D) ^d
3′	77.50 (Sm)	77.88 (Sm)	133.36 (S)
4′	27.41 (Qm)	28.01 (Qm)	18.20 (Q) ^b
5′	27.41 (Qm)	28.01 (Qm)	25.92 (Q) ^e
1″	72.81 (T)	72.06 (T)	72.37 (T)
2″	119.13 (Dm)	119.99 (Dm)	119.84 (D) ^d
3″	139.95 (Sm)	138.52 (Sm)	138.95 (S)
4″	18.05 (Qm)	18.11 (Qm)	18.02 (Q) ^b
5″	25.74 (Qm)	25.82 (Qm)	25.85 (Q) ^e
			• • •

^{*a*} Numberings of the above compounds correspond to those of compound (6). ^{*b-e*} Assignments may be reversed.

spectrum of (2). This suggested that compound (2) contains a 2,2-dimethylchromene skeleton [*cf.* arugosin E (1)], instead of the 3-methylbut-2-enyl group present in (6). All the above results and the consideration of the co-occurrence of emericellin (6)⁸ support the structure of cycloisoemericellin as that given in (2).

In order to determine the structure of cycloisoemericellin (2) conclusively, long-range proton selective decoupling experiments in the ¹³C n.m.r. spectrum were performed. The carbon signals at $\delta_{\rm C}$ 118.84 (Dq, C-5) and 140.50 (Sq, C-6) were changed into a simple doublet and singlet, respectively, by selective irradiation of the protons at δ 2.42 assigned to the aromatic methyl group. The multiplicity of the carbon at $\delta_{\rm C}$ 152.38 (Sm) was also changed in the above conditions. On the other hand, when the protons at δ 7.25 and 6.86, which were assigned to the aromatic protons coupled with each other, were irradiated at the same time, the five carbon signals at $\delta_{\rm C}$ 154.39 (Sdd, C-1), 116.91 (Sddd, C-2), 156.42 (Sdd, C-4a), 122.86 (Sd, C-9a), and 121.61 (Dd, C-1') were changed into a doublet, double-doublet, singlet, singlet, and doublet, respectively. The above results confirmed the structure of cycloisoemericellin as (2) and the assignments of the ¹³C n.m.r. signals in Table 6.

It is interesting that the arugosins, which were originally isolated from the fungi of the *A. nidulans* group: *Emericella* variecolor and *Emericella rugulosa*, were also isolated from *A. silvaticus* and *E. striata*. According to Fennell and Raper,⁹ *A. silvaticus* is a non-ascosporic form which appears to be intermediate between the *A. versicolor* and *A. nidulans* groups. However, the chemotaxonomic profile of *A. silvaticus*, *E. variecolor*, *E. rugulosa*, and *E. striata* is strikingly suggestive of their placements in the same group, viz. the *A. nidulans* group.

Experimental

M.p.s were determined on a Yanagimoto micro-melting point apparatus and are uncorrected. E.i. m.s. were recorded on a

JEOL JMS-D 300 spectrometer. U.v. spectra and i.r. spectra were recorded on a Hitachi 124 spectrophotometer and a Hitachi 215 spectrophotometer, respectively. ¹H (99.60 MHz) N.m.r. spectra were recorded on a JEOL JNM-FX 100 spectrometer, while ¹³C (100.43 MHz) n.m.r. spectra were recorded with a JEOL JNM-GX 400 spectrometer, using tetramethylsilane as internal standard. The coupling patterns are indicated as follows: singlet = S or s, doublet = D or d, triplet = T or t, and quartet = Q or q. Capital letters refer to the pattern resulting from directly bonded coupling $({}^{1}J_{C,H})$. Column chromatography was performed using Kieselgel 60 (Art. 7734; Merck). Low pressure liquid chromatography (l.p.l.c.) was performed on a Chemco Low-Prep pump 81-M-2 and glass column (200 \times 10- or 20-mm) packed with silica gel CQ-3 (30-50 µ; Wako). T.l.c. was conducted on precoated Kieselgel 60 F₂₅₄ (Art 5715; Merck). Spots on t.l.c. were detected by their absorption under u.v. light, and/or by iodine vapour.

Isolation of Arugosins from Aspergillus silvaticus.—Aspergillus silvaticus, strain IFO 8173, was cultivated at 27 °C for 21 days in Czapek-Dox medium, using 340 Roux flasks containing 250 ml of the above medium in each flask. The dried mycelia (470 g) were extracted with chloroform. The organic layer was dried (Na₂SO₄) and concentrated under reduced pressure to give the chloroform extract (37 g). Chromatography on silica gel with benzene, followed by repeated purification by l.p.l.c. with benzene gave anhydroarugosin (40 mg), and arugosins A (3) and B (4) (4.0 g). Further elution with benzene-acetone (50:1) followed by repeated purification by l.p.l.c. with benzene gave dithiosilvatin² and arugosin E (1) (30 mg) successively.

Arugosin E (1) was obtained as yellow needles or prisms (from benzene), m.p. 159—161 °C (Found: C, 70.8; H, 6.2. $C_{25}H_{26}O_6$ requires C, 71.07; H, 6.20%); m/z 422 (M^+ , 16%), 354 [($M - C_5H_8$)⁺, 53], 339 (73), 325 (100), 321 (25), 270 (22), 187 (37), 161 (63), and 69 [(C_5H_9)⁺, 55]; λ_{max} (MeOH) 214 (log ε 4.42), 273 (4.26), and 348 nm (3.64); v_{max} (MeOH) 214 (log ε 4.42), 273 (4.26), and 348 nm (3.64); v_{max} (MeOH) 214 (log ε 4.42), 273 (4.26), and 348 nm (3.64); v_{max} (MeOH) 214 (log ε 4.42), 273 (4.26), and 348 nm (3.64); v_{max} (MeOH) 214 (log ε 4.42), 273 (4.26), and 348 nm (3.64); v_{max} (MeOH) 214 (log ε 4.42), 273 (4.26), and 348 nm (3.64); v_{max} (MeOH) 214 (log ε 4.42), 273 (4.26), and 348 nm (3.64); v_{max} (MeOH) 214 (log ε 4.42), 273 (4.26), and 348 nm (3.64); v_{max} (MeOH) 214 (log ε 4.42), 273 (4.26), and 348 nm (3.64); v_{max} (MeOH) 214 (log ε 4.42), 273 (4.26), and 15 (20 cm⁻¹; δ_H (CDCl₃) 0.94 (6 H, s, aliphatic Me \times 2), 1.63 (3 H, s, olefinic Me), 1.77 (3 H, s, olefinic Me), 2.36 (3 H, s, ArMe), 4.39 (2 H, d, J 7.3 Hz, OCH_2CH=), 5.26 (1 H, d, J 9.8 Hz, CH=CH), 5.40 (1 H, br s, PhOH), 5.51 (1 H, br t, J 7.3 Hz, OCH_2CH=), 6.12 (1 H, d, J 9.8 Hz, CH=CH), 6.49 (1 H, d, J 8.6 Hz, ArH), 6.99 (1 H, s, ArH), 7.01 (1 H, d, J 8.6 Hz, ArH), 10.19 (1 H, s, CHO), and 12.43 (1 H, s, PhOH). The ¹³C n.m.r. signals are summarized in Table 6.

Isolation of Cycloisoemericellin (2) from Emericella striata.— The benzene-acetone (50:1) fraction obtained from column chromatography of the acetone extract (26 g) of the mycelium (660 g) of *Emericella striata*, strain 80-NE-22, was crystallized from acetone to give ergosterol,⁸ and the remaining residue was chromatographed on silica gel with hexane-ethyl acetate (20:1, v/v) followed by l.p.l.c. with cyclohexane as solvent to give arugosin A (3) and B (4) (40 mg) and with hexane-ethyl acetate (10:1, v/v) followed by recrystallization from ether-hexane to give cycloisoemericellin (2) (92 mg).

Cycloisoemericellin (2) was obtained as yellow needles, m.p. 136—138 °C (Found: C, 74.15; H, 6.75. $C_{25}H_{26}O_5$ requires C, 73.86; H, 6.45%); *m/z* 406 (*M*⁺, 7%), 337 [(*M* - C₅H₉)⁺, 100], 319 (32), 305 (78), 295 (31), 267 (20), 255 (38), and 69 [(C₅H₉)⁺, 26]; λ_{max} .(MeOH) 238 (log ε 4.35), 280 (4.62), 317sh (3.91), and 380 nm (3.74); ν_{max} .(KBr) 3 450 (OH), 1 640, and 1 610 cm⁻¹; δ_{H} (CDCl₃), 1.57 (6 H, s, aliphatic Me × 2), 1.73 (3 H, br s, olefinic Me), 1.81 (3 H, br s, olefinic Me), 2.42 (3 H, d, *J* 0.7 Hz, ArMe), 4.45 (2 H, br d, *J* 7.3 Hz, OCH₂CH=), 4.98 (1 H, s, OH), 5.02 (2 H, br s, CH₂OH), 5.61 (1 H, br t, *J* 7.3 Hz, OCH₂CH=), 5.64 (1 H, d, *J* 9.8 Hz, CH=CH), 6.33 (1 H, d, *J* 9.8 Hz, CH=CH), 6.86 (1 H, d, *J* 8.3 Hz, ArH), 7.21 (1 H, br s, ArH), and 7.25 (1 H,

d, J 8.3 Hz, ArH). The 13 C n.m.r. signals are summarized in Table 6.

Oxidation of Arugosins A (3) and B (4) with DDQ.—DDQ (180 mg) was added to a solution of arugosin A (3) [arugosin B (4)] (240 mg) in dry benzene (20 ml), and the mixture was heated under reflux for 30 min. The precipitate was filtered off, and the reaction mixture was washed with water several times then evaporated under reduced pressure. The residue was repeatedly purified by 1.p.l.c. with benzene to give arugosin E (1) (90 mg), which was identified by its i.r. and ¹H n.m.r. spectra and t.l.c. and h.p.l.c. behaviour.

Structure Determination of Arugosin E (1) by X-Ray Diffraction.—Diffraction intensities were collected from a crystal of dimensions $0.75 \times 0.50 \times 0.30$ mm of arugosin E (1) on a Rigaku AFC-5 FOS four-circle diffractometer. Of the total 3 392 reflections (complete for $7^{\circ} \le 2\theta \le 120^{\circ}$), 2 888 satisfied the criterion $F \ge 3\sigma(F)$ and only these were used in the solution and refinement of the structure.

Crystal Data.—C₂₅H₂₆O₆, M = 422.5, monoclinic, a = 9.045(14), b = 17.282(13), c = 14.606(15) Å, $\beta = 92.84(11)^\circ$, V = 2 280.3 Å³, Z = 4, $D_c = 1.231$ g cm⁻³, F(000) = 896, space group $P2_1/c$, Cu- K_{α} X-radiation (graphite monochromator), $\lambda = 1.5418$ Å.

Structure Solution and Refinement.—The structure was solved by direct methods using MULTAN 80¹⁰ and in the final refinement by block-matrix least-squares method; anisotropic thermal parameters were used for all non-hydrogen atoms. The contribution of hydrogen atoms was ignored. The refinement converged to R 0.092 and R_w 0.102. The data were not corrected for the effects of absorption. Positional parameters are shown in Table 2, and bond lengths and angles are summarized in Tables 3 and 4, respectively. A list of anisotropic and equivalent thermal parameters is available on request from the Cambridge Crystallographic Data Centre.*

Acknowledgements

We are grateful to Professor M. Yamazaki of the Faculty of Pharmaceutical Sciences, Chiba University, and Dr. S. Udagawa of National Institute of Hygienic Sciences, for helpful discussions. We thank Professor J. Shoji and Dr. Y. Ida of Showa University for mass measurement. We are also grateful to Mrs. T. Ogata, Mrs. M. Yuyama, and Miss T. Tanaka of this university for elemental analyses, and for n.m.r. and mass measurements, respectively.

* See para. 5.6.3 in 'Instructions for Authors (1988),' J. Chem. Soc., Perkin Trans. 1, 1988, Issue 1.

References

- 1 K. Nozawa, S. Udagawa, S. Nakajima, and K. Kawai, *Chem. Pharm. Bull.*, 1985, 35, 3460.
- 2 N. Kawahara, K. Nozawa, S. Nakajima, and K. Kawai, J. Chem. Soc., Perkin Trans. 1, 1987, 2099.
- 3 J. A. Ballantine, D. J. Francis, C. H. Hassall, and J. L. C. Wright, J. Chem. Soc. C, 1970, 1175.
- 4 J. A. Ballatine, V. Ferrito, C. H. Hassall, and M. L. Jenkins, J. Chem. Soc., Perkin Trans. 1, 1973, 1825.
- 5 K. K. Chexal, J. S. E. Holker, T. J. Simpson, and K. Young, J. Chem. Soc., Perkin Trans. 1, 1975, 543.
- 6 K. K. Chexal, J. S. E. Holker, and T. J. Simpson, J. Chem. Soc., Perkin Trans. 1, 1975, 549.
- 7 S. M. Anand and A. C. Jain, Tetrahedron, 1972, 28, 987.

- 8 H. Seya, K. Nozawa, S. Nakajima, K. Kawai, and S. Udagawa, J. Chem. Soc., Perkin Trans. 1, 1986, 109. 9 D. I. Fennell and K. B. Raper, Mycologia, 1955, **47**, 68.
- D. H. Folken and R. D. Raper, Mytologia, 1995, 47, 66.
 P. Main, S. J. Fiske, S. E. Hull, L. Lessinger, G. Germain, J. P. Declercq, and M. M. Woolfson, 'MULTAN 80. A System of

Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data,' 1980, Universities of York, England, and Louvain, Belgium.

Received 1st May 1987; Paper 7/783